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## THE <sup>13</sup>C-NUCLEAR MAGNETIC RESONANCE SPECTRA OF PHYSOSTIGMINE AND RELATED COMPOUNDS\*

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The medicinally valuable alkaloid, physostigmine (1a), has been long established as the major basic component of the seeds of *Physostigma venenosum* (Calabar beans)[1]. More recently the alkaloids physovenine (1b), eseramine (1c) and N<sub>8</sub>-norphysostigmine (1d) have also been extracted as minor basic components of these seeds [2,3].

As well as containing ring systems which are likely to be biogenetically derived from tryptophan, these alkaloids all possess *N*-methylcarbamyl groups. Although the *O*-carbamyl group is present in the structure of novobiocin [4], and the *N*-carbamyl group is present in the biogenetic intermediates citrulline [5] and carbamyl

(2)

aspartate [6], its presence in the bases of *Physostigma* venenosum seeds, where it is probably biogenetically derived from carbamyl phosphate (cf. refs. [5] and [6]) is as yet unique in higher plant alkaloids. As a prerequisite to a biosynthetic study on this group of alkaloids and as a useful aid in the structural investigation of further bases isolated from this source, we have examined the <sup>13</sup>C-NMR spectra of the above four alkaloids and report below (Table 1) their full assignment.

The carbon assignments are based on proton noise decoupled, off-resonance decoupled and proton coupled spectra together with comparison with literature data [7] and the <sup>13</sup>C-NMR spectrum of the analogous indoline (2) [8]. The correlations are generally very close, but useful diagnostic shifts are seen in the series, particularly at C-2 where a downfield shift of 14.1 ppm for physovenine (1b) and an upfield shift of 7.5 ppm for eseramine (1c) are observed, relative to the C-2 value in physostigmine. Similarly C-8a exhibits an upfield shift of 7.8 and 8.9 ppm in N<sub>8</sub>-norphysostigmine (1d) and eseramine (1c) respectively, and a downfield shift of 6.6 ppm in physovenine (1b). The 8-methyl group also shows an upfield shift of 7.2 and 4.8 ppm respectively in physovenine (1b) and eseramine (1c).

## EXPERIMENTAL

The <sup>13</sup>C-NMR spectra were recorded at 20 MHz on a Varian CFT-20 spectrometer in the quoted solvent. Chemical shifts are given in ppm relative to TMS as internal standard. The concentrations of the solutions used were 10-20% and the spectra were recorded at approximately 30%.

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Table 1. <sup>13</sup>C-NMR spectral assignments in *Physostigma venenosum* alkaloids

Compound (solvent)	Assignment (ppm)														
	I	2	3	3a	3a-Me	3b	4*	5	6*	7	7a	8-CH <sub>3</sub>	8a	$\mathrm{CH_3NH^\dagger}$	COt
Physostigmine (1a) (CHCl <sub>3</sub> )‡	36.9	53.2	40,7	52.6	27.2	137.4	116.1	149.3	120.4	106.5	143.3	38.4	98.1	27.5	156.3
N <sub>8</sub> -Norphysostigmine (1d) (CHCl <sub>3</sub> )	37.0	52.5	40.7	53.7	26.9	137.8	116.5	146.9	120.5	109.0	[44.0]		90.3	27.9	156.3
Eseramine (1c) (DMSO)	CO157.7 Me 23.3	45.7	38.5	50.4	26.9	135.1	116.0	147,4	120.7	105.8	142.8	33.6	89.2	26.9	155.6
Physovenine (1b) (CHCl <sub>3</sub> )		67.3	41.6	52.3	24.6	135.2	116.5	147.9	120.8	105.5	143.0	31.2	104.7	27.7	156.3
Compound (2) (CHCl <sub>3</sub> )		*****		42.7	12.0 22,7	140.0	115.5	149.2	120.0	107.4	143.8	34.1	72.9 Me 25.5	27.4	156.3

<sup>\*</sup> These assignments are interchangeable; † compare MeNHCOOC<sub>2</sub>H<sub>5</sub> (Me--27.4; CO--157.8 ppm) [7]; ‡ coupling constants:  $J_1$  (CH-7)--160.1;  $J_1$  (CH-6/4)--160.8 and 158.8;  $J_1$  (CH-3a-Me)--127;  $J_1$  (CH-8-Me)--134;  $J_1$  (CH-8a)--155;  $J_1$  (CH-2)--142;  $J_1$  (CH-3)--125;  $J_1$  (CH-MeNH)--138;  $J_3$  (CH-4/6)--3.9 and 4.9 Hz.

<sup>\*</sup> Part 11 in the series "Alkaloids of *Physostigma veneno-sum*". For Part 10 see Dale F. J. and Robinson B. (1970) *J. Pharm. Pharmacol.* **22**, 889.

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## ALKALOIDS OF THREE ASPIDOSPERMA SPECIES

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**Key Word Index**—Aspidosperma formosanum, A. campus-belus, A. desmanthum; Apocynaceae; isolation; olivacine; uleine; 3-epiuleine; 1,13-dihydro-13-hydroxyuleine; aspidocarpine; lichexanthone; phthalimide; aspidoalbine.

Plants and sources. Aspidosperma formosanum A. P. Duarte (Formosa, Goiás, Brazil, 1965; APD herbarium register 9387); A. campus-belus A. P. Duarte (Campos Belos, Goiás, 1965, APD register 9481); A. desmanthum Benth. ex Müll.—Arg. (IPEAN, Belém, Pará, Brazil, 1965, APD register 9798). Previous work: None; A. formosanum is systematically close to A. dasycarpon [1] (Series Tomentosa); A. campusnbelus to A. nigricans [2] (Series Pyricolla); A. desmanthum to A. exalatum [3], A. spruceanum [2a], and A. album [4] (Series Nobile).

Bark. Hot continuous EtOH extraction followed by conen gave in each case about 10% syrupy extract. This was macerated with 2N HOAc, filtered, and divided into standard fractions [5] (letter code; method of obtention; percent of extract in the case of A. formosanum, A. campus-belus, and A. desmanthum, respectively); A, C<sub>6</sub>H<sub>6</sub> extraction of the aq. HOAc solution, 0.87, 2.1, 1.47; B, CHCl<sub>3</sub> extraction of the solution after neutralization with HCO<sup>-</sup><sub>3</sub>, 2.53, 7.1, 2.2; D, CHCl<sub>3</sub> extraction after basification to pH 13 with NaOH, 1.75, 0.8, 0.83.

In the preliminary testing of the various extracts, olivacine (1) was noted as the principal base in fraction B of A. campus-belus, and a small quantity obtained by direct crystallization from MeOH was compared satisfactorily with material from A. nigricans [2].

In large-scale work, the following compounds were isolated (plant; fraction(s), isolation methods, compound name and structure number, yield based on dried bark,

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mp, other relevant data for characterization, confirmation of identity):

- A. desmanthum. C, direct crystallization from MeOH, aspidoalbine (2), 0.05%,  $174-175^{\circ}$  (lit.  $174-177^{\circ}$  [4],  $168^{\circ}$  [2a]); MS showing possible impurity of the *N*-acetyl analogue (3) at m/e 414, but not evident in the NMR; comparison of spectral data [4].
- A. formosanum. (1) A,B,C; direct crystallization from MeOH, or basic Al<sub>2</sub>O<sub>3</sub> III eluting with hexane-C<sub>6</sub>H<sub>6</sub> (1:1) to  $C_6H_6$ , or with toluene to toluene-EtOAc (1:1), or with hexane-CH<sub>2</sub>Cl<sub>2</sub> (4:1) to CH<sub>2</sub>Cl<sub>2</sub>, or Si gel eluting with EtOAc-MeOH (9:1); uleine (4); 0.64%; 72-78°, but highly variable (known to be poorly crystalline and solvated and show wide melting ranges [2,6]);  $[\alpha]_D^{27}$ +20° (CHCl<sub>3</sub>; c 0.94),  $\lambda_{\text{max}}^{\text{MeOH}}$  nm 213, 307, 315 (log  $\epsilon$  4.38, 4.28, 4.24),  $\nu_{\text{max}}^{\text{CHCl}_3}$  cm<sup>-1</sup> 3534m, 2941s, 1767w, 1637m, 1621m, 1460s, 1445s, 1314s, 1148m, 1125m, 1098m, 1047m, 1007m, 977w, 935w, 911w, 873s, 839m, NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  8.72 (1Hs, eliminated with D<sub>2</sub>O; NH), 7.40–6.80 (4Hm; ArH), 5.18 and 4.84 (2 × 1Hs; =CH<sub>2</sub>), 3.95 (1Hd, J 3 Hz; C-4), 2.16 (3Hs; N-Me), 1.04 (2Hq, J 6 Hz; C-14), and 0.76 (3Ht, J 6 Hz; C-15), MS M 266 (100%) and fragmentation as published [7], comparison with an authentic sample (B. Gilbert). Significance: the large amount of this alkaloid present, its relatively facile isolation, and its unusual and suggestive 1-methylene-4-aminotetrahydrocarbazole structure, have led us to explore chemical transformations into analogues of antischistosomicidal drugs (preazaquinone methides), which will be reported upon in another Journal.
- (2) A,B, after preliminary crystallization of uleine; neutral Al<sub>2</sub>O<sub>3</sub> I eluting with hexane–C<sub>6</sub>H<sub>6</sub> (4:1); 3-epiuleine (5); 0.013%; amorphous; UV identical to that of uleine,  $v_{\rm max}^{\rm CHCl_3}$  cm<sup>-1</sup> 3521*m*, 2941*s*, 1767*w*, 1637*m*, 1621*m*, 1460*s*, 1445*s*, 1314*s*, 1140*m*, 1125*m*, 1101*m*, 1043*m*, 1010*m*, 978*m*, 952*w*, 910*w*, 870*s*, 823*m*, NMR MHz, CDCl<sub>3</sub>)  $\delta$  7.98